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STUDY OF PERFORMANCE PARAMETERS AND EXHAUST GAS POLLUTION OF NATURAL GAS FUELED SPARK IGNITION ENGINE

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BY

MD. SYED ALI HOLLA

A Thesis submitted to the Department of Mechanical Engineering .in partial fulfillment for the degree

of

"ASTER OF SCIENCE IN "ECHANICAL ENGINEERING

BANGLADESII UNIVERSITY OF ENGINEERING & TECHNOLOGY, DHAKA, BANGLADESH

STUDY OF PERFORMANCE PARAMETERS AND EXHAUST GAS POLLUTION OF NATURAL GAS FUELED SPARK IGNITION ENGINE

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DECLARATION

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This is here by declared that neither this thesis nor any part there of has been submitted or is being concurrently submitted any where else for the award of any degree or diploma or for any publication.

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ABSTRACT

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Among engine, largest partially the liquid fossil fuel (gasoline and diesel). many alternative fuels for internal combustion natural gas can be considered as the single alternative fuel which can substitute totally or

This research work was intended to study the performance characteristics and exhaust gas emissions (CO,HC,C02, 02) of a two cylinder diesel engine converted to spark ignition engine to run on 100 percent compressed natural gas (CNG).

In this experiment, Tartarini gas regulator of Itally, gas air mixer, electronic ignition system, spark plugs, etc. were fitted. Compressed natural gas (CNG) cylinder containing gas at 200-210 kg/cm2 was used as a fuel source.

From the experimental result, it is seen that natural fueled spark ignition engine can deliver nearly equal output of its diesel counterpart (unconverted engine). gas

The natural gas fueled spark ignition engine was found to be more efficient at the ignition timing of 15° BTDC (before top dead . centre) wheh the Brake Specific Energy Consumption (BSEC) of 13.22 Mj/kwhr and Brake Specific Fuel Consumption (BSFC) of 293.82 gm/kwhr and Brake Thermal Efficiency (BThEff) of 27.23% were found at full load operation.

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The ignition engine is found to be 120°C to 180°C above than that of a diesel engine and a petrol engine of comparative power range. exhaust gas temperature of natural gas fueled spark

During the test-CO,HC,CO₂ and O₂ were measured. NO_x is usually measured for diesel engine's where the formation of NOx is high due to high combustion temperature and high compression ratio. exhaust gas pollutants

The harmful exhaust gas pollutant like carbon monoxide(CO) which restricts oxygen transfer in blood hemoglobin is found nil at the air fuel ratio above 17.48:1 for natural gas fueled spark ignition engine indicating that natural gas is a clean fuel.

It is also found that at air fuel ratio below 17.48:1,the exhaust gas pollutant (CO) is traceable in considerable amount depending upon load condition and richness of the air fuel mixture. $\alpha \approx$

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The author also appreciates the sincere co-operation of the General Manager, Senior Manager, Plant Manager and other Technical Staff of Compressed Natural Gas (CNG) Plant Ltd, Dhaka, Bangladesh from whom instruments, equipments, materials and technical informations were collected.

The extensive co-operation of the General Manager, Additional Chief Engineer, Mr. A. Majid Chowdhury, Mechanical Engineer, Dr. A. Hossain Mollah and other Technical Staff of Bangladesh Diesel.Plant (BDP) Ltd,is highly appreciated specially for providing the two cylinder diesel engine along with other two for modification and research work and also for arrangement and repairing of Hydraulic Dynamometer of BUET in their factory.

The author thanks Mr. Andrew Cambell of Liquid Fuel Management Group Ltd., Newzealand who modified and converted this two cylinder diesel engine to eNG fueled spark ignition engine in the Heat Engine Laboratory of BUET as a part of research work .managed by Energy Division of the Government of Bangladesh under the financial assistance of the British Government.

Lastly the author thanks the staff of Heat Engine Laboratory, Boiler Laboratory, Machine Shop, Welding Shop, Carpentry Shop, Fluid Mechanics Laboratory and other staff who have participated directly or indirectly in this research work.

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$(viii)$

LIST OF FIGURES

 $\bar{\mathcal{L}}$

 $\ddot{}$

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3.4 The variation of brake specific energy consumption (BSEC) , brake thermal efficiency (B.Th.Eff.), exhaust gas temperature (Exht. temp.) and air fuel ratio (AFR) with respect to brake power (Kw) at a constant speed of 1500 rpm and at an ignition timing of 6° BTDC.

 $\sim 10^6$

 $\omega_{\rm{max}}$.

44

 (x)

- 3.5 The variation of brake specific energy consumption (BSEC) with respect to brake power (Kw) at different ignition timings from 6° BTDC to 15° BTDC at a constant speed of 1500 rpm. 45
- 3.6 The variation of brake thermal efficiency (B.Th.Eff.) with respect to brake power (Kw) at different ignition timings from 6° BTDC to 15°BTDC at a constant speed of 1500 rpm. 46

47

49

50

51

52

 α and α .

- 3.7 The variation of exhaust gas temperature temperature with respect to brake power (Kw) at different ignition timings from 6°BTDC to 15°BDTC at a constant speed of 1500 rpm.
- 3.8 The **variation** of **maximum** exhaust gas 48 respect to the different ignition timings from 6°BTDC to 15° BTDC at constant speed of 1500 rpm.
- 3.9 The variation of maximum load carrying capacity of the engine (Kw) with respect to the different ignition timings from 6°BTDC to 15°BTDC at a constant speed of 1500 rpm.
- 3.10 The variation of exhaust gas emissions
(CO,HC,CO₂ and O₂) with respect to bral with respect to brake power (Kw) at a constant speed of 1500 rpm and at an ignition timing of 15°BTDC.
- 3.11 The variation of exhaust gas emissions (CO,HC,C02 and 02) with respect to brake power (Kw) at a constant speed of 1500 rpm and at an ignition timing of 12°BTDC.
- 3. 12 The variation of exhaust gas emissions (CO,HC,C02 and 02) with respect to brake power (Kw) at a constant speed of 1500 rpm and at an ignition timing of 9°BTDC.
- 3.13 The variation of exhaust gas emissions (CO,HC,C02 and 02) with respect to brake power (Kw) at a constant speed of 1500 rpm and at an ignition timing of 6°BTDC. 53
- 3.14 The variation of the percentage of carbon monoxide pollution $(x^{\dagger}$ CO v/v) with respect to different air fuel ratio at a constant speed of 1500 rpm.
- 3.15 The variation of hydrocarbon pollution (HC level in ppm) with respect to brake power (Kw) at different ignition timing from 6°BTDC to 15°BTDC at a constant speed of 1500 rpm. 55
- 3.16 The variation of the percentage of 56 oxygen **emission** (% 02 v/v) with respect to different **air** fuel ratio at the ignition timing of 15° BTDC and at a constant speed of 1500 rpm.
- 3.17 The **variation** of brake thermal efficiency 57 with respect to equivalence ratio at the ignition timing of 15°BTDC and at a constant speed of 1500 **rpm.**

54

LIST OF TABLES

Table No

 $\bar{\beta}$

Page No.

(xiii)

LIST OF APPENDICES

 $\mathcal{A}^{\mathcal{A}}$

Description

 $\bar{\epsilon}$

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 \mathbb{R}^2

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 $\hat{\mathcal{A}}$

Page No.

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CHAPTER -1

INTRODUCTION :

Fuel crisis and its subsequent effect was acutely felt in Western countries' like U.K., West Germany, France, Italy, U.S.A. along with other developed countries in early seventies when the price of liquid fuel rose abnormally high. Since then developed countries took keen interest in finding out the alternative in the major fuel depending sectors. The necessity of alternative fuel in S.I (spark ignition) and C.I (compression ignition) engine also got importance in many countries. Subsequently lot of
research and success have been achieved in countries like and success have been achieved in countries like Italy, U.S.S.R., New Zealand, Canada,' Denmark,' U.S.A., Japan, France, etc. Tartarini of Italy has developed different type of gas regulators, mixer, cylinder and other gas controlling units .which are commercially used world wide. Japan and France have developed gas engine of different ranges upto 750 Kw.

The alternative fuels so far sorted out are Compressed Natural Gas (CNG), Liquified Petroleum Gas (LPG), Methanol, Ethanol, Bio-mas, etc. Each alternative fuel has got advantages, disadvantages and own economics, but compressed natural gas has top position among them for using in I.C. engine because of its more advantages and less disadvantage like-good combustion with less engine noises and vibrations, less exhaust gas pollutants and satisfactory engine performances.

In Bangladesh, research on natural gas fueled S.I. engine is limited with Petro Bangla, BUET, and Bangladesh Diesel plant CNG Plant under Petro Bangla converted a few number of gasoline fueled S.I. engine to natural gas fueled S.I. engine in collaboration with Tartarini of Italy and TNO of Netharland. TNO is working as a consultant and Tartarini as the supplier of CNG conversion kits. CNG plant under Petro Bangla is simply taking up the work of converting gasoline fueled S.I. vehicle to natural gas fueled S.I. vehicle with the conversion kit imported from .Tartarini of Italy but no experimental data are available there.

Since Bangladesh has got sufficient gas reserve against 100% dependence on imported petroleum fuel, so the use of CNG in S.I. engine, in C.I. engine and converted S.I. engine can play a vital role towards the saving of foreign courrency and dependence on foreign coutnries for fuel.

Bangladesh Diesel Plant (BDP) Ltd manufactures 2 cylinder diesel engine locally. If this diesel engines can be converted to dedicated gas. it will generate economic and social' values and develope experties locally.

The present research work has the following objectives -

- (i) To study the effect of ignition timing on the performanc of natural gas fueled spark ignition engine at different loads.
- (ii) To study the overall performance characteristics of the engine at a constant speed of 1500 rpm with variable load test.
- (iii) To study the exhaust gas emissions ($CO₁HC$, $CO₂$ and $O₂$) of natural gas fueled spark ignition engine for correlating the exhaust gas pollution / emissions with engine performance characteristics.
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CIIAPTER-2

LITERATURE REVIEW

Nagosh and Samaga (20) optimised a biogas fueled 5.1. Engine. In their test result it is reported that engine performance in general more satisfactory at advanced ignition (35°BTOC) and at larger spark plug gap (0.45mm). The tested engine showed a minimum requirement of biogas at the rate of 0.7 m³ /bhp-hr. The biogas contained 59% methane and other combustible and 41% COz.

Karim and Ali (16) tested a single cylinder spark ignition research engine fueled with natural gas . Their result showed that in the exhaust, oxides of nitrogen (NOx) always decreased with the retarded spark ignition timing. The emission of CO increased with decreasing mixture temperature and also with increasing compression ratio.

Cambell (6) conducted experiments on the use of compressed natural gas (CNG) in spark ignition engine and in dual fuel engine. He converted a single cylinder diesel engine (Duetz 210 $D, 6.7$ Kw, air cooled, comp. ratio 17:1) to a natural gas fueled S.I. engine reducing compression ratio to 12:1. From his test result of single cylinder 5.1. engine, he reported that exhaust tempertature decreased ignificantly with ignition timimg 'advanced. An increase in ignition advance from 2° BTDC to 15° BTDC ... resulted in decrease in exhaust temperature from 702°C to 620°C at 2000 rpm at rated load. It was also mentioned that exhaust temperature in natural gas fueled engine was overall
high compared to diesel fuel operation. Exhaust emission diesel fuel operation. Exhaust emission was found to follow typical spark engine trends.

Obert (23) explained the performance characteristics for a constant speed load test of I.C. engine . It is mentioned that with the increase of load, brake specific fuel consumption decreased due to increase of mechanical efficiency at high load.

Lyon. Howland' and Loom (17) proposed the technique of controlling smoke and gaseous emissions from a diesel engine, using liquid petroleum gas as alternative fuel . They suggested that direct injection of diesel responded excellently to dual fueling and this method could be used to control smoke.

Baluswamy (2) worked on the LPG dual fuel engine using 70% diesel and 30% LPG and compared this result with 100% diesel
operation. He concluded from his experiment that LPG dual fuel He concluded from his experiment that LPG dual fuel operation always led to lower S.F.C. values and also found that smoke was greatly reduced. From an examination of the various heat release record, they suggestd that dual fuel com .bustion generally appeared to undergo two distinct phases, the first phase was associated mainly with the

combustion of pilot and associated gaseous fuel, while the second phase was dependent on the concentration and quality of the gaseous fuel employed. The heat release analysis of lean mixture operation showed agreement with other experimental evidence that comperatively inefficient combustion under such condition was due to mainly the inability of the gaseous charges to supplemen effectively the heat release of the first phase . The use of larger pilots or higher initial temperature could reduce this tendency.

Murty (22) investigated kinematic NO (oxide of nitrogen) inside the cylinder of S.I. engine working on gasoline wate mixture . He mentioned that formation of NO in the engin cylinder depended not only on the various parameters but also on the character of combustion process which in term depended on the design and other factors. In the investigation result it it is seen that Nitric Oxide (NO) increases from 250 ppm to 400 ppm for natural aspirated engine but Nitric Oxide (NO) increases from 500 ppm to 650 ppm to supercharged hydrogen engine when inlet
temperature increased from 290 deg K to 320 deg K. It was temperature increased from 290 deg K to 320 deg K. It was further observed that as a result of lean air fuel mixture and high peak combustion temperature NO emission was high.

Newton, Steeds and Garretts (21,) mentioned that unburnt hydrocarbon in the exhaust originated from the quench layer on the walls of combustion chamber, a few hundredths millimeter thick in which the flame was extinguished by cooling. Therefore surface /volume ratio must be kept small as possible and design should be free from crevices and corners where quench might occur.

Beg (4). studied the effect of mixing on the performance of a single cylinder dual-fuel diesel engine. The experiment was conducted with different shape of air-gas mixers, entry
lengths, and different percentage of diesel gas consunption. lengths, and different percentage of diesel gas consunption. From the experimental result it is reported that for a particular load, as the percentage of diesel consumption decreased, the equivalent Specific Fuel Consumption increased and the rate of increase was found to be higher at lower percentage of diesel consumption. This increase was more prominent at lower loads than higher loads. At higher loads, equivalent SFC was very close to 100 percent diesel operation though the gas flow rate increased. From the comparison of equivalent SFC for different air-gas mixers and entry length, it was reported that with 60 percent diesel consumption, equivalent SFC decreased by 3.8 percent for cross-shape, 3.7 percent for Lshape, and 3.5 percent for L-shape (direction of L opposite to air flow) mixer. This was due to better mixing of air and gas before entry of the engine.

Bari (3) also conducted an experiment on the study of the effect of dual fuel on some performance parameters of a single cylinder small diesel engine. It is reported that though the rated power of the original diesel engine was 5 Hp but the same engine gave maximum 3.15 Hp with dual fuel operation. Three types of "mixers (Direct, L-shape and Cross-shape) were used in the experiment and among those the Cross-shape mixer was found to be more efficient. It was also reported that at 75% of full load and 60% diesel consumption, the SFC was about 1% less than the 100% diesel fuel operation. It was concluded that the combustion in dual fuel engine mainly depended on three factors were homogenity of the mixture, flame speed and load.

CHAPTER - 3

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METHODOLOGY :

Diesel engine does not need much modification for running it with dual fuel mode. In dual fuel operation of a diesel engine, the CNG or natural gas is supplied at low pressure in the. intake manifold of the diesel engine and diesel is injected in limited quantity as a pilot fuel for starting the combustion of the air fuel mixture in side the cylinder.

But the same diesel engine needs a lot of modification for operating it by hundred percent dedicated CNG fueL The engine has to be fitted with ignition system, fuel induction system along with the reduction of the compression ratio. In this experiment the compression of the diesel engine were also reduced from 17:1 to 12:1 and the compression ignition system was replaced by the installation of electronic ignition system.

The details steps and procedure for modification of the diesel engine to hundred precent CNG fueled spark ignition engine is described in appendix A-18..

$CHAPTER - 4$

DESCRIPTION OF THE EXPERIMENTAL SET-UP :

A schematic diagram of the experimental set-up has been shown in fig.l.l.l. The experiment was set up with the following equipments and measuring devices etc.

- a. A two cylinder Duetz Diesel Engine (12.68Kw) converted it to CNG fueled $S.I.$ Engine. The details of the engine specifications are listed in the appendix table $A-2.$ specifications are listed in the appendix table
- b. Hydraulic Dynamometer for measuring engine load during experiment.
- c. Air supply system including its air measuring and controlling devices as shown in schematic figure no. 1.2.1
- d. Fuel (CNG) supply system including its gas weighing device (electronic weighing scale), pressure regulator, gas flow control valves, pressure pipes etc. as shown in shcematic figure no. 1.2.1..
- e. 270 Crypton (NDIR) Emission (CO,IIC,COz, 021 Analyser with an additional hose pipe, water separator etc.
- f. Manometers group for measuring different pressures (Orifice pressure, exhaust gas depression, intake manifold pressure, air inlet depression).
- g. Digital thermocouple groups for measuring different temperatures (air box temperature, lube oil temperature, air fuel mixture temperature at the intake manifold, exhaust gas temperature, cylinder head temperature, dynamometer exit water temperature).
- h. 'Mixer.(gas carburator) for proper mixing of fuel and air.
- i. Digital tachometer for measuring engine rpm.
- j. A sling phychrometer for measuring dry bulb temperature and wet bulb temperature.
- k. A Barometer for measuring atmospheric pressure.

The diesel engine was modified to S.I. Engine by reducing its compression ratio from 17:1 to 12:1 and fitting an electronic ignition system on it. Photographs of the engine fitted on the test bench are shown in fig. 2.1. 1. and 2.1.2. The important technical details of the original diesel engine and converted CNG fueled S.I. Engine are listed in appendix table A-I and A-2.

The schematic diagram of air flow system is shown in fig. 1.2.1. The original air filter was removed from the engine and an air flow measuring device was fitted with the intake manifold of engine. The air flow and its measuring device consisted of an air drum (actually two nos. empty drums fabricated and welded to single drum), an orifice meter with corner tapping, air hose pipe (dia 12.7cm), air shut-off valve inclined manometer, as shown in f ig. 1.1.2. The air shut-off valve w ²⁵ installed before air hose The air shut-off valve $w = s$ installed before air hose (i.e. at the exit of air from drum) for controlling intake manifold air pressure and its flow as and when necessary.

The 'gas (CNG) supply system consisted of two nos. CNG cylinders connected in parallel by pressure pipe and gas control valves, so that gas (CNG) could be supplied to engine from any one of these two cylinders as and when necessary. One CNG cylinder was for supply and storage of CNG gas, called storage cylinder. The other CNG cylinder called service cylinder was fitted on a_c weighing electronic scale under the test bench which supplied eNG during the period of the experimental work. This weighing or service cylinder was refilled from the CNG of storage cylinder through control valves before starting the experiment.

.Electronic weighing scale was used for taking weight of service cylinder before, and after the experimental observation from where fuel flow rate was calculated. Digital stop watch has been used for measuring observation'period during the engine test.

High pressure gas (CNG) passed from CNG cylinder to gas regulator (pressure reducer) where high pressure CNG (200-210 bars) was reduced to its line pressure of approximately 1 kg/cm² from where the gas passed into the mixer and it is then mixed with air to feed a homogenious mixture of gas and air to the engine cylinder.

The engine was installed on an engine test bench and its out put shaft was coupled to a hydraulic dynamometer with necessary fittings and cpuplings. The fittings were fabricated in the machine shop and welding shop of BUET.

The exhaust gas from the exhaust silencer was passed outside the engine room through a fabricated G.I. pipe (10.16 cm). Provission was also made for collection of exhaust gas sample from exhaust line to gas analyser. A separate pot or wate separator was also fabricated and installed on the line of exhaust gas sample, so that it could separate water from the exhaust sample before its entry into the analyser.

A group of manometers were installed to measure different pressures like exhaust gas depression, intake air pressure, **manifold vacuum, air orifice pressure etc.**

A sling physhrometer was installed for measuring dry bulb and wet bulb temperature near the engine.

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$CHAPTER - 5$

PROCEDURE

The experiment was conducted in the following steps:

6.1 Checking of CNG Reserve and Pressure in CNG Cylinders:

Before starting the experiment, .it should be confirmed that CNG stock and its pressure is sufficient for taking at least one complete set of observation. As such before starting the engine, CNG pressure in the storage cylinder and in the service cylinder was checked individually by opening the gas control valve and observing the corresponding gas pressure from the pressure gauge fitted at the inlet line of the gas regulator.

In our case, 100 kg/cm² pressure of natural gas in storage cylinder and in service cylinder was sufficient for taking a complete set of robservation.

5.2 Warming up and making the Crypton 270 Emission Analyser Ready for use:

It takes nearly 10 minutes to make the emission analyser ready for use. As such before starting the engine, the analyser should be brought up into operation mode as per instructions of operating manual of 270 Crypton Emission Analyser. The filter was particularly inspected for water deposit as water in the
filter gives wrong reading. In rainy, season, water data it is filter gives wrong reading. In rainy season, water deposit in
the water seperator was found in so much quantity that at seperator was found in so much quantity that at every time of engine starting, sufficient quantity of water needed to be removed.

After those preliminary checkings the power was supplied from the switch board to the gas analyser through a voltage stabliser (1 Kw). Then the emission analyser was allowed for its self calibration for bringing it into operation mode.

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5.3 Checking Engine and Dynamometer :

Before starting the" engine, oil level and its leakage were checked up thoroughly and oil was added or refilled as and when necessary. Moreover battery connections were also checked up and tightened if necessary.

Oil level in dynamometer bearings were also checked up and oil was added if necessary.

Obstructions to engine rotation and dynamometer rotation were also checked up and removed if any.

More over different measuring devices were thoroughly inspected before starting the engine.

5.4 Engine Operation and Experimental Observation :

After ensuring the natural gas supply from the storage CNG cylinder, the engine was started with the help of starting key fitted on the switch board fixed on the engine mounting

;' Engine was allowed to warm up for 5-10 minutes and then load and speed were increased slowly. Engine was fixed at 1500 rpm and dynamometer's load was increased to the value for maximum carrying capacity of the engine. This was done by simul taneously increasing the water supply to the dynamometer through the water regulating valve from the storage water tank to dynamometer inlet and increasing the entry of air fuel mixture through the throttle valve under the gas carburator, keeping the $-e$ ngine rpm at 1500 rpm.

Experimental data in this position was recorded . The dynamometer's load on the engine was reduced by lkg step from the maximum load carrying capaity to its half value and in each step data were recorded without changing the ignition timing.

During experiment the engine was set at the ignition timing of 15°BTDC and at the constant speed of 1500 rpm. The engine load was varied at the step of 1 Kg load (approximately) and at each step experimental data for Engine Brake Power, Air Flow Rate, Fuel Flow Rate, Temperatures and Pressures for study of engine performance characteristics and exhaust gas emissions (CO,HC,COz, Ozl were also recorded.

Later on, the ignition timing was varied (reduced) at the interval of 1.50°. At each ignition timing interval, engine load was varied as mentioned above at a constant speed of 1500 rpm.

The emission analyser gave the direct reading of x CO v/v, x $CO₂$ v/v, % $O₂$ v/v and HC level in ppm.

Power, rated power, specific fuel consumption, specific energy consumption, thermal efficiency, equivalence ratio, etc.were calculated from the experimental data. Sampl calculation is shown in Appendix A-17.

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CHAPTER - 6

RESULT AND DISCUSSION

6.1 GENERAL:

The diesel engine (2 cylinders) converted to hundred percent compresssed natural gas fueled spark ignition engine was tested at a constant speed of 1500 rpm with variable loads, for the study of the performance characteristics and exhaust gas emissions at different ignition timings from 6°BTDC (before top dead centre) to 150BTDC (before top dead centre) at 1.5° interval.

The fig. no. 3.1 to fig. no. 3.4 are drawn from the test data for the ignition timings of 15°BTDC, 12°BTDC. 9°BTDC and 6°BTDC for study of the performance characteristics at different ignition timings. Each parameter (BSEC, BThEff, Exht Temp.) of the performance characteristics at different ignition timings (150BTDC, 120BTDC,90BTDC and 60BTDC) has been drawn and compared to fig. no. 3.5, 3.6 and fig. no. 3.7. Moreover fig. no. 3.8 and fig. no. 3.9 show variations of maximum brake power and exhaust gas temperature at different ignition timings. Also fig. no. 3.10 to fig. no. 3.15 are presented from the test data for the study of exhaust gas emissions.

6.2 INVESTIGATION OF PERFORMANCE CHARACTERISTICS :

From the test result and figures, the observations regarding brake specific energy consumption, brake thermal efficiency, minimum brake specific energy consumption, exhaust gas temperature, maximum brake power, the effect of air fuel ratio on the engine brake thermal efficiency, percentage of carbon monoxide (CO) emission, carbon dioxide (C02) emission, oxygen (02) emission and level of hydrocarbon (HC) emission are presented here.

6.2.1 Brake Specific Energy Consumption :

It is seen from the fig. no. 3.1 to 3.4 that the brake specific energy consumption decreases with the increase of load upto 65% to 75% of the full load operation and again the same (BSEC) increases with the increase of loads till its full load operations in all the tests under different ignition timings from 60BTDC to 15°BTDC.

6.2.2 Brake Thermal Efficiency :

It is also found that brake thermal efficiency increases
the increase of loads till its peak value and then falls with the increase of loads till its peak value and then again in all the different ignitions timings as shown in fig. no. 3.1 to 3.4. Again the brake thermal efficiency of different ignition timings has been compared in fig. no. 3.6 from where it is seen that the engine is more efficient at the ignition timing of 15°BTDC.

6.2.3 Minimum Brake Specific Energy Consumption (Mj/kw-hr):

Brake specific energy consumptions at different ignition timings from 60BTDC to 15°BTDC have been compared in fig. no. 3.5 from where it is observed that the minimum specific energy consumption decreases with the advance of the ignition timing.

The Minimum Brake Specific Energy Consumption decreases from 14.97 Mj/kw-hr to 10.92 Mj/kw-hr with the corresponding increase of Maximum Brake Thermal Efficiency from 24.04% to 32.94% when
the ignition timing is advanced from 69BTDC to 159BTDC. This the ignition timing is advanced from 6°BTDC to 15°BTDC. result is compared with the previous test result (6) in the table below.

6.2.4 Exhaust Temperature

It is also observed from fig. no. 3.7 that exhaust temperature decreases with the advance of ignition timing. It is
further observed from fig. no. 3.8 that maximum exhaust further observed from fig. no. 3.8 that maximum exhaust
temperature decreases from 698°c to 554°c when the ignition temperature decreases from 698°c to 554°c when the timing was advanced from 6°BTDC to 15°BTDC.

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But while comparing this exhaust temperature with that of a typical diesel engine or a gasoline engine, it is seen that exhaust gas temperature of CNG fueled engine is higher than of diesel and gasoline fueled engine. that

In general the exhaust temperature of diesel engine and gasoline engine is within the temperature range of 37000 to 410oc at full load operation but the exhaust temperature can be reduced to 584 oc by adjusting (advancing) the ignition timing for CNG fueled spark ignition engine at full load operation.

6.2.5 Maximum Brake Power :

It is observed from fig. 3.9 that the maximum brake power increased with the advance of ignition timing. The brake power increased from 10.09 Kw to 12.68 Kw when ignition timing was advanced from 6°BTDC to 15°BTDC. The maximum power is obtained at the ignition timing of 15°BTDC.

6.2.6 The Effect of Air Fuel Ratio on the Engine Brake Thermal Efficiency :

The effect of air fuel ratio on the engine brake thermal efficiency showed typical relation. Substantial increase of engine brake thermal efficiency could be obtained by proper air fuel mixure, as found from fig. no. 3.17. The complete burning at proper air fuel mixutre causes the increase of brake tharmal efficiency.

, 6.3 INVESTIGATION OF THE EXHAUST GAS EMISSIONS:

Though the exhaust gas pollutants are CO , NO_x , SO_2 , HC particles but our investigation was limited to analyse t'he percentage of CO, C02, 02 and HC level for natural gas fueled S.I. Engine. There were no facility for measuring NO_x in our test and moreover NO_x meter is generally used for measuring emission of diesel engine where the formation of the oxides of nitrogen (NO_x) is likely to be high due to high combustion temperature inside the cylinder for its high compressor ratio.

6.3.1 CO Emission

At air fuel ratio above 18.84:1, CO emission is found nil in all the different ignition timing from 6°BTDC to 15°BTDC as apparent from fig. no. 3.14.

The carbon monoxide (CO) emission is found to increase with rich fuel operation and it is true for all the different ignition timing. 'The percentage of CO is found to ,decrease with the lean air fuel mixture as seen from the above figures no. 3.14. Percentage of Carbon monoxide emission found nil even at 85% of full load operation and it raises to 0.6% v/v when the output (kw) is 93% of its full load and percentage of CO further rises to 1.9% at its full load operation. While comparing the CO emission of CNG fueled spark ignition engine with that of admissible limits of CO pollutants in USA legislative or accepted limit of exhaust gas pollutants in B.E.C. and Japan, it is seen that the harmful CO level is in safe level and much lower than that of gasoline or diesel fuel operation as shown in the table below :

So it found that the harmful CO emission is well below the accepted limits of USA, Japan, and EEC. Moreover CO emission can be
reduced to zero by leaping air fuel ratio above 17.48:1 . at 15° reduced to zero by leaning air fuel ratio above 17.48:1, at 15°
rund, when the lead garrung capacity is about 85% of full load. BTDC when the load carryng capacity is about 85% of full load.
The test procedure of exhaust gas pollutants in USA is different The test procedure of exhaust gas pollutants in USA is different
from laboratory test, since in USA, exhaust gas pollutants, are from laboratory test, since in USA, exhaust gas pollutants measured by collecting them in bag while the vehicle is operated inside and outside of town at different speed and modes on a chassis mounted dynamometer. But Toyota Company recommended measure CO when engine run at hot condition while the vehicle is kept "stationary for its test.

6.3.2 HC Emission :

HC is the unburnt hydrocarbon originating from the quench
setthe unlig of combustion, a few bundredths millimeter in layer on the walls of combustion, a few hundredths millimeter which the flame is extinguished by cooling. The other cause, of the presence of hydrocarbon in the exhaust is the misfiring when the unburnt hydrocarbon of the full cylinder is escaped into the exhaust gas. The level of hydrocarbon is found high when the fuel
in the mixture is high as apparent from fig. no. 3.10 to 3.1 . in the mixture is high as apparent from fig. no. 3.10 At too lean mixture the level of hydrocarbon is found to rise again. From the experimental data, the fig. no. 3.10 shows that the level of HC is traced even at lean mixture. This is mainly due to insufficient fuel in the mixture to start firing.

Through there are several proposals of the HC level limit in U.S.A. from 1966 to 1975 but no level limit of HC is available for world wide or for their national level application. But the
manufacturer, of Crynton 270, Emission, Analyser, of U.K., has manufacturer of Crypton 270 Emission Analyser of U.K. has
recommended the level of HC upto 400 ppm as normal. But in our recommended the level of HC upto 400 ppm as normal. But in case it is 160 ppm at full load.

So also accepted limit. it is found that HC level in CNG fueled S.I. Engine is safe side and comparatively low and it only 40% of the

6.3.3 C02 Emission

C02 in the exhaust gas indicate the combustion efficiency and it is the equilibrium product of combustion. CO_2 -curve in the fig. no. 3.9 to 3.13 is found to follow typical CO_2 - curve trends
of gasoling engine, The level of CO_2 is found to raise, at its of gasoling engine. The level of $CO₂$ is found to raise at peak value at the air fuel ratio arround 17.48: 1 and comes

down again. At optimum air fuel ratio (17.48:1),the percentage of $CO₂$ becomes high because of complete combustion of the air fuel mixture. At rich mixture, percentage of C02 is lower because of incomplete combustion of air fuel mixture. Again at too lean air fuel mixture, percentage of C02 is also low because of the fact that excess air containing only 0.4% C02 bring the overall percentage of $CO₂$ in the exhaust gas to a low level.

While comparing the percentage of $CO₂$ in the experimental work with that of the value of the ultimate analysis of natural gas in the table $A-4$, it is found that the percentage of the CO_2 of the experimental result is slightly less than that of the theoritical value as shown in the table below :

The above 0.52 % discrepency in the percentage of $CO₂$ probably for the leakage of air through the exhaust gas joints. is most manifold

6.3.4 02- Emission :

Level of O_2 emission in the exhaust gas has been shown in fig. no. 3.10 to 3.13 for different ignition timing. Moreover percentage of $oxygen(0₂)$ with respect to air fuel ratio in the exhaust gas emission is shown in- fig.no. 3.16 for the ignition timing of 15 degree before top dead centre and it is found that with rich mixture, percentage of O_2 is low because of the fact that most of the oxygen is burnt in the combustion process. Again percentage of oxygen is found to decrease with leaner fuel-air mixture because most of the oxygen remains unburnt in leaner mixture.

CHAPTER -7

CONCLUSIONS

7.1 General Conclusion:

From the experimental result it is seen that the overall performance characteristics are competitive and quite satisfactory in natural gas fueled spark ignition engine. Moreover the exhaust gas pollutants (CO/HC) are also very low and below the admissible limits of the exhaust gas pollutants for natural gas fueled spark ignition engine.

The formation of the oxides of nitrogen(NOx) in natural gas fueled vehicle can also be controlled by installing exhaust gas
recinculation system (EGR) with natural gas fueled rucicu consesser
recirculation system (EGR) with natural gas fueled vehicles/engines if necessary. Duetz 2FL 912 diesel engine can be adopted to run on dedicated CNG with appropriate modifications.

7.2 Future Recommendation:

Though our research was limited with variable load test at constant speed, in future research, the natural gas fueled spark ignition engine should be optimized at a variable speed load test along with the study of the side effects like the durability of engine oil viscosity and wear and tear of engine parts like piston. rings, cylinder liners, bearings etc. for modified 2FL912 engine.

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Fig. 1.1.1 Schematic Diagram of The Experimental Set-up.

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- 14. Hydraulic dynamometer
- 15. Funnel
- 16. Water regulating valve
- 17~ Reserve water tank
- 18. Digital tachometer
- 19. Digital tachometer stand
- 20. Exhaust pipe
- 21. Water seperator
- 22, Crypton 270 emission **analyser**
- 23. Voltage stablize
- 24. Step down tranforme
- 25. Power supply board

 F_2 g. 1.1.2 Schematic Diagram of Air and Fuel Flow System.

1. 2. 3. 4. 5 • 6. 7 . 8. Power/load valve 9 • 10 11. Shutter 12. Air drum CNG cylinder (storage) CNG cylinder (service) CNG cylinder shut-off valve 16. Flexible plastic pipe High pressure hose High pressure pipe Gas shut-off valve Pressure regulator Mixer (gas carburator) Air hose

*13. Air drum stand

14. Air orifice

15. Inclined manometer

17. Throttle valve

18. Flange with nut- bolt

19. Two cylinder engine unit 20. Piston

21. Sensor with hanger of weightronic scale

22. Weightronic scale

23. Adopter

24. Voltage stabliser

25. Power plug of pannel board

Fig. 1.2 Schematic Diagram of Fuel Flow System.

1. Fuel cylinders 2. Cylider shut-off valve 3. Master shut-off 4, Refilling point 5. High pressure line

6. Regulator ~ 10

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- 7. Gas line
- 8. Vacumm line
- 9. Water from radiator
- 10. Water return line
- 11. Carburator
- 12. Petrol shut-off valve
- 13. Changeover switch

 $26₂$

27

Fig. 2.1.1 Photograph of the converted eNG Fueled .Spark Ignition Engine at the time of its installation on the engine test bench-

Fig. 2.1.2 Potograph of the Engine, Dynamometer, Engine Test Bench along with other accessories where the engine load **is** varied by controlling the water regulating valve for supply of water from storage water tank to dynamometer.

Fig- 2.2 Photograph of the Crypton 270 Emission Analyser, lKw stac Voltage Stablizer and Power Distribution Box**29**

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Fig. 2.3 Photograph of Crypton 270 Emission
Stac Voltage Stablizer during warm up period. Analyser and

Fig. 2.4 Photograph of the eNG Fueled **S.I.** Engine showing

switch board for starting the engine **31**

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V. \hat{y} .

Fig. 2.5 Photograph of the CNG Fueled S.I. Engine
showing its speed adjustment.

Fig. 2.6 Photograph of Weightronic Scale, Stop Watch,
Adopter, CNG-Service-Cylinder-and-Pressure-Gauges.

Fig- 2.7 Photograph of Inclined Manometer intake air. for measuring

34

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Fig. 2.8 Photograph of Sling Psychrometer.

KEY TO DIAGRAM

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1 C.H.G. cylinders, connected to the main pipe-fine.
Inside these cylinders the gas pressure resches 2800 **PS1**

2. Cylinder etcp valves.

. Main food plps...
. Filling valva, which allows the filling of the cylinders vithout having to move them.

without having to move them.
6. Pitler for the natural pas to the pressure repulator,
6. Pressure pauge to control the natural pas pressure
5. On G, aolenoid valve with eléctric primer, This de-

 $7 \text{ C} \text{ N G}$, actioned value with sidelific primer. This de-
when earlies out three important functions: strops the
gestion the statistics pressure regulator, avery time
the ignition hay in withdrawn from the dash board particular movement, thus facilitating starting when
the angina is cold

the angina is cold

p Pressure regulator: Decompresses the natural pas

from the pressure 2800 PSI in the cylinders in the

aure from the passes the gas to the miser in the

guentlife required by the engine.

p Stow-runnin

10. Pipe, connecting the eacuum to the inteke mani-

fold.
13. Pipe, connecting the pressure regulator to the

the first site.
The first production water to the pressure regulator.
The first control is the mater from the pressure regular 13. Pipe, circulating water from the pressure regula-

tor,
14. Mixer, fixed on the carburator,
18. Mestmum flow edjusting acrew

Meximum flow adjusting acrew.

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Certurator
Gasolina sciencid valve. Stops the Row of Puet.

TF Georgine schemuld values Steps the Boston Christian Companies of the High phase of the High Apple Steps of the particle procedure of the particle procedure of the particle procedure of the Steps Christian Christian Chr

- ice zwei
- 22 Spritton key on deshtioerd
- 24 Golf
- 24 Golf
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Photograph of CNG Pressure Pequlator and CNG Flow $Fig. 2.9$ System including it Electrical Wiring and other accessories.

Petrol Pump

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Fig. 10 Photograph of CNG Conversion Eits Installation
Gasoline Fueled Vehicles. in

267 CO/HC Analyser 270 CO/HC/CO2/O2 Analyser

 $-2\cdot 11$ Photograph of 270 Crypton - Emissions (CO/HC/CO₂/O₂)

Fig. 2.12 Photograph
Emissions Analyser. Front Pannel of 270 Crypton o f

Fig. 2.13 Thotograph of Emission Analyser fitted with
during observation of emissions from running vehicle. vehicle.

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Brake power (Kw)

Fig. 3.1 The variation of brake specific energy consumption (BSEC), brake thermal efficiency (B.Th.Eff), exhaust gas temperature (Exht.temp.) and air fuel ratio (AFR) with respect to brake power (Kw) at a constant speed of 1500 rpm and at an ignition timing of 15° BTDC.

Brake power (KW)

of brake specific energy Fig. 3.2 The variation consumption (BSEC), brake thermal fficiency (B.Th.Eff), exhaust gas temperature (Exht.temp.) and air fuel ratio (AFR) with respect to brake power (Kw) at a constant speed of 1500 rpm and at an ignition timing of 12° BTDC.

Brake power (KW)

variation of brake specific energy Fig. 3.3 The consumption (BSEC), brake thermal efficiency (B.Th.Eff), temperature (Exht. temp.) and air fuel
with respect to brake power (Kw) at a gas exhaust (AFR) ratio and at an ignition timing constant speed of 1500 rpm of 9° BTDC.

43

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of brake specific energy The variation consumption (BSEC), brake thermal efficiency (B.Th.Eff), gas temperature (Exht.temp.) and air fuel ratio (AFR) with respect to brake power (Kw) at a exhaust constant speed of 1500 rpm and at an ignition timing of $6°$ BTDC.

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Fig. 3.5 The variation of brake specific energy consumption (BSEC) with respect to brake power (Kw) at different ignition timing from 6°BTDC to 150BTDC at a constant speed of 1500 rpm.

variation of exhaust gas temperature The 3.7 Fig. with respect to brake power(Kw) at different ignition timing from 6°BTDC to 15°BDTC at a constant speed of 1500 грт.

 $Fig.3.8$ The variation \circ f maximum exhaust gas temperature with respect to the different ignition timings from 6°BTDC to 15° BTDC at constant speed of 1500 rpm.

48

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Spark ignition timing (deg. BTDC)

The variation of maximum load Fig.3.9 carrying capacity the engine (Kw) with respect to the \circ f different ignition timings from 6°BTDC to 15°BTDC at a constant speed of 1500 \widetilde{rpm} .

Brake power (K W)

Fig. 3.10 The variation of exhaust gas emissions $(CO, HC, CO_2$ and O_2) with respect to brake power (Kw) with respect to brake power (Kw) at a constant speed of 1500 rpm and,at an ignition timing of 15°BTDC.

Fig. 3.11 The variation of exhaust gas emissions $(CO, HC, CO₂$ and $O₂$) with respect to brake power (Kw) at a constant speed of 1500 rpm and at an ignition timing of 12°BTDC,

The variation of exhaust gas emissions Fig. 3.12 (CO, HC, CO2 and O2) with respect to brake power(Kw) at a constant speed of 1500 rpm and at an ignition timing of 9°BTDC.

The variation of exhaust gas emissions Fig. 3.13 (CO, HC, CO2 and O2) with respect to brake power(Kw) at a constant speed of 1500 rpm and at an ignition timing of 6°BTDC.

Fig. 3.14 The variation of monoxide pollution (%CO v/v) $_{\mathtt{air}}$ fuel ratio at a constant the percentage of carbon with respect to different speed of 1500 rpm.

variation of hydrocarbon pollution (HC Fig. 3.15 The ppm) with respect to brake power (Kw) at
ignition timing from 6°BTDC to 15°BTDC at a level in different constant speed of 1500 rpm.

Fig. 3.16 The variation of the percentage of Oxygen
emission (% O2 v/v) with respect to different air fuel at the ignition timing of 15°BTDC and at a ratio constant speed of 1500 rpm.

56

The variation of brake thermal efficiency with respect to equivalence ratio at the ignition timing of $Fig. 3.17$ 15°BTDC and at a constant speed of 1500 rpm.

57

TEST RESULT AT THE IGNITION TIMING OF 15°BTDC.

Power(S)*- Power at Standard condition, W_{fe}-Standard fuel flow rate.

Table no. 1

58

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TEST RESULT AT THE IGNITION TIMING OF 13.5° BTDC.

	Sl. Item No. No.						Obs.1 Obs.2 Obs.3 Obs.4 Obs.5 Obs.6 Obs.7		Unit of item.
							1 Power(S) 12.14 11.13 9.90 8.72 7.50 5.79 4.50		КW
	2 W_{f}		1.11 0.84 0.75 0.70 0.61						0.58 0.50 gm/sec.
3.	Wa		$16.54 - 14.57 - 13.07 - 12.08 - 11.05$						10.34 9.20 gm/sec.
\blacktriangleleft	AFR						14.86 17.29 17.38 17.21 18.06 17.78 18.36		
5.			BSEC 14.81 12.23 12.27 13.01				13.18 16.28 18.00		Mj/Kwhr
6.							BThEff 24.30 29.44 29.33 27.80 27.37 22.18 20.00		x
			7 Ex-Tem 617 619 634			621 603	594 - 1	587 -	$\mathfrak{o}^{\mathbf{C}}$
							8 E.R. 1.15 0.99 0.99 0.99 0.95 0.96 0.93		
9.	%CO.		2.4 0.9 0.0 0.0 0.0				0.0 0.0		V/V
10	HC.		80.0 60.0		0.0 0.0	20.0	20.0 20.0		ppm.
11			$x \text{CO}_2$ 6.0 7.0				8.0 8.0 6.0 6.0	5.0	\sqrt{v}
							$12 \quad \& 0_2$ 11.2 11.5 11.8 12.2 12.6 13.1 13.5		$-V/V$
	$13 - 5$	75.0		75.0 75.0 72.0		$72 - 0$	72.0 72.0		γ.
14	$P_{\bf a}$			749.58					mmHg

Table No. 1.2

59

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Table no. 1.3

	Sl. Item No. No.								\mathbf{r} . Qbs.1 Obs.2 Obs.3 Obs.4 Obs.5 Obs.6 Obs.7 Unit of item.
	1 Power(S) 11.11 10.57 10.01 8.90 7.79 6.57							5.79	ΚW
	2 Wes								0.90 1.19 1.18 0.85 0.68 0.54 0.36 gm/sec.
3.	$W_{\rm m}$								14.24 19.43 19.33 13.99 11.57 10.08 8.94 gm/sec.
4.	AFR 15.77 16.27 16.33 16.40 16.95 18.60 24.77								
5.	BSEC								13.13 18.24 19.10 15.47 14.14 13.12 10.07 Mj/Kwhr
								6 BThEff: 27.43 19.97 19.02 23.28 25.46 27.04 35.74 %	
						627 600	584 -	575 —	۰C
	8 E.R. 1.09 1.06 1.05 1.05 1.01 0.92 0.69								
	9 2 0	3.80	3.50 2.60					0.40 0.00 0.00 0.00	\sqrt{v}
	10 HC	160	135 140			20 0.0	20.0	20.0	ppm
								$11 \quad \text{\%CO}_{22}$ 6.0 6.0 6.0 7.0 8.0 6.0 6.0 v/v	
12	%0 _z	10.4	10.4	10.6	10.9	11.8	12.6	12.9	v/v
13 ₁	Ø×	72.5	$69.0 -$	- 69.0			69.0 69.0 69.0	69.0	٧,
14	P_{α}	\sim			752.33				mmHg

TEST RESULT AT THE IGNITION TIMING OF 10.5° BTDC.

Table no. 1.4

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 \mathcal{L}_{max}

TEST RESULT AT THE IGNITION TIMING OF 9° BTDC.

 \overline{a}

Table no 1.5

62 \bar{A}

	Sl. Item No. No.						$0b5.1$ $0b5.2$ $0b5.3$ $0b5.4$ $0b5.5$ $0b5.6$	Unit of item.
	1 Power(S) 10.92 9.94 8.89 7.78 6.66 5.55							a kw
	2 $W_{\tau, \alpha}$						0.97 0.89 0.75 0.71 0.68 0.60	gm/sec.
	3 Wa				15.50 15.38 13.75 12.28 11.84		11.00	gm/sec .
	4 AFR				16.00 17.26 18.32 17.28	17.39	$16 - 16$	
								5 BSEC = 14.39 14.51 13.67 14.78 16.54 17.51 = Mj/Kwhr
	7 BThEff 25.02 28.82 26.34 22.42 21.76 20.37							У.
	8 Ex.Tem 644 669 667 655 636 633							۰c
	9 E.R.				1.11 0.99 0.94 0.98 0.99 0.94			
	9 %CO			$2.2 \t 0.9 \t 0.0$	$Q \bullet Q$	0.0000000		V/V
	10 HC		150 140	140	80.	20 -	$0 - 0$	ppm
	11 % CO_{2}	- 8	8.	9.	$7 -$	7 ¹	5.	v/v
	$12 \times 0_2$	10.7	10.9	$11 - 3$	$11 - 7$	12.0	12.3	V / V
13	Ø.	79	80.5	79.5	80	30	80	χ
14	$P_{\rm m}$				750,19			mmHg

TEST RESULT AT THE IGNITION TIMING OF 7.5° BTDC.

Table no. 1.6

TEST RESULT AT THE IGNITION TIMING OF 6°BTDC.

Table no. 1.7

APPENDIX TABLE A-I

 $\overline{(i)}$ for Deutz pump (ii) for Bosh pump

APPENDIX TABLE A-2

Duetz 2 1 500 *'r* l2 : 1 Electronic ignition system $CR-9E$ (for NGK) $G-$ 58 (for Champion) 19mm 0.3 to 0.5 mm 75 , degree *-----r--------* ._---~----!-----------------------------~---- SPECIFICATION OF CONVERTED ENGINE (CNG FUELED S.I. ENGINE) Engine No. of cylinder Rpm (tested) Compression ratio Ignition system Spark plug (type) Spark plug size Spark plug gap Dual angle **---**

٠j.

COMPOSITION OF NATURAL GAS BY WEIGHT (9) • Methane Ethane Propane 94.3502226 2.56423 1.09133 1.08266 Butane Nitrogen Carbondioxide 0.2545388 0.67257

APPENDIX TABLE A-4

PROPERTIES OF NATURAL GAS(24)

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APPENDIX TABLE A-5

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 $\mathbf{C}^{\mathbf{t}}$

TABLE FOR LIMIT OF FLAMIBILITY OF GASES IN AIR AT S.T.P. WITH UPWARD PROPOSAL (13)

APPENDIX TABLE A-6

IGNITION TEMPERATURE OF COMMON FUEL GAS IN AIR AT ATMOSPHERIC PRESSURE (13)

APPENDIX TABLE A-7

TABLE OF MAXIMUM FLAME SPEED AND CORRESPONDING MIXTURE OF GAS IN AIR (13)

APPENDIX TABLE A-8

SPECIFICATION OF CNG CYLINDER USED IN CNG FUELED VEHICLE IN BANGLADESH (*)

* - CNG Ltd, Bangladesh

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68

APPENDIX TABLE A-9

TRADE BALANCE AND PETROLEUM IMPORT OF BANGLADESH (24) (In Million U.S. Dollar)

 $\hat{\mathcal{A}}_1$

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 $\hat{\mathbf{Q}}$

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APPENDIX TABLE $A - 10$

(Natural gas reserve in Bangladesh)

13 (thirteen) gas feilds have been discovered in the country upto now. Out of these gas feilds only Titas, Habigonj, Haripur (Sylbet) and Chatak have been devoloped and at present producing gas from il (eleven wells), five at Titas, two at Habigonj, two at Haripur, one at Chatak and one at Lailashtil. producing at an verage of 212 macft per day.

 \mathbf{r}

Note: tncf- Trillion Cubic Feet - 1012

nnef- 10° cft, bbl- American barrel, btu- British Thermal Unit Source- Petrobangla.

APPENDIX TABLE $A-11$ BS 5514 / ISO 3046) (FROM

Annex F

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Determination of water vapour pressure .

The water vapour pressure (e_1, p_{33}) is given in the table below in units of kPa for different values of the air temperature i_s in e_1 Celsius and telative humidity e_{\pm} . \hat{z} .

APPENDIX TABLE A-12 (FROM BS 5514 / ISO 3046)

\mathcal{L}_{CFT} , \mathbf{x},\mathbf{t}

"Determination of dry air pressure ratio

he dry air pressure ratio $\left(\frac{p_x - ac_x p_y}{p_x - ac_x p_y}\right)$ used in formula (3) is given in the tante below for the value of $\chi_{\rm eff}$ and $\chi_{\rm eff}$ and the different cases of total barometric cressive (p_s) and water vanour precsure follows. the water vapour pressure is not known it can be obtained from the air temperature and relative humidity by the use of annex 4

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APPENDIX TABLE $A - 13$ BS 5514 / ISO 3046) $(\texttt{From}$

73

Amnex D

Determination of the ratio of indicated power (k)

" Formula (3) or (4) can be written as $(k \geq (k_1)$ ") $(R_2V_2(R_3))$

where
$$
R_3 = \frac{P_3 - 3P_3 P_3}{F_1 - 9P_2 P_3}
$$
 or
$$
\frac{P_3}{P_{10}}
$$

$$
R_2 = \frac{T_1}{T_1} \text{ or } \frac{T_{10}}{T_2}
$$

$$
R_3 = \frac{T_{\text{c}}}{T_{\text{c}}}
$$

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and $y_1 = m$ $y_2 = q$ $y_3 = q$

The value of $R = \frac{p_1 - a c_1 p_{11}}{p_1 - a o_1 p_{11}}$ can be obtained from annex E and other values of R can be calculated.

The values of m, n, a are obtained from table 1 .

The table below then gives values of Rz for known ratios R and known factors y

The value of k is then obtained by multiplying together the appropriate values of R^1 .

 $\begin{bmatrix} 1 \\ 1 \end{bmatrix}$

$A - 14$ **APPENDIX**

Annex C

 74

The table below gives values of the fuel consumption adjustment factor (β) for known values of the ratio of indicated power (k) and mechenical efficiency (n_m) .

The value of k can be determined from annex D .

The value of η_m is stated by the manufacturer (see clause 16, note 4).

APPENDIX TABLE A-15 (From BS 5514 / ISO 3046)

 \mathcal{L}_{max} , and \mathcal{L}_{max} , and

 $\mathbf{A}^{\mathbf{c}}$

Annex B

 $\mathcal{L}^{\mathcal{A}}$

Determination of the power adjustment factor (a)

the table below gives values of the nower adjustment factor follow kalons on the ratio of indicated power (4), and mechanicat Hiciency (n_m).

the value of k can be determined from annex D. .

The value of η_m is stated by the manufacturer (see clause 10, note 41,

APPENDIX TABLE $A-16$ $(From$ BS $5514 / ISO$ 3046)

 76

(Eumerical values for power acquatment .)

NOTE - The factors and exponents given in table 1 have been established by tests on a number of engines to be generally representative and shall be used in the absence of any other specific information; for example in formula reference D, for an engine with the charge air cooled by engine jacket water, the value for exponent q could be zero. At present, they apply only to the types of engines specified but table 1 will be extended to include other types when sufficient data are available.

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APPENDIX A-17

SAMPLE CALCULATION

A-17.1.1 AIR FLOW RATE CALCULATUION

Air flow rate is calculated using sample data of appendix table A-17.11. From B.S. 1042, air flow rate is calculated by the following formula-Air flow rate = 0.01252 CZe Ed² (ha f_a)^{1/2} Where, W_a = Air flow rate in Kg/hr. C = Constant of orifice. d = Diameter of orifice. h_a = Pressure difference across the orifice meter in mmH₂O f_a = Density of the air at the orifice meter in Kg/m³ Z ,e and E -approximately equal to one. $0.01252 \times 0.596x1000x d^2(h_a f_a)^{1/2}$ W_a gs⁻¹ = ----------------60x60 $= 0.002073 d^2(h_a f_a)^{1/2}$ P_a $752.18x13.6x62.4$ Again, $\mathbf{f}_a = ---$
RT_a $25.4x12x53.3x(460+91.4)$ $= 0.0713$ lb/ft³ 0.0713 1 ⁼ **----------** ^x **-------** Kg/m³ 0.0283 $= 1.1419$ Kg/m³ $= 1.142$ Kg/m³ Therefore , $W_a = 0.002073 \times (30.02)^2$ (ha fa)^{1/2} $= 1.868188(h_a f_a)^{1/2} = 16.306$ for fa=1.142 Kg/m $h_a = 66.709$ mmH₂₀

 $= 16.306$ gm/sec. and $t_a = 33^{\circ}C$

Where β is d/D

A-17.3 CALCULATION OF FUEL FLOW RATE

Initial wt. of service cylinder Final weight of service cylinder $w_1 = 25.30$ Kg, measured by weightronic scale. $w_2 = 24.880$ Kg (at the end of observation) Time of observation $t = 337.25 \text{ sec.}$,

measured by digital tachometer

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Hence fuel flow rate Wf (25.30-24.880)xlOOO = ---------------- 337.25

= 1.038gm/sec

A-17.3 AIR FUEL RATIO

Air flow rate 16.30G Air Fuel Ratio(AFR) = $\frac{15.709}{1.038}$ Fuel flow rate

A-17.4 POWER CALCULATION

BHP= NLxO.OOl for hydraulic dynamometer used in the experiment.

Brake horse power (metric horse power) Where, BHP ω, Speed in rpm. Indicated load on dynamometer scale in Kg. L $\frac{1}{2}$

0.001 Constant of the dynamometer.

Hence BHP=1500x11.2x0.001 = 16.8 for N= 1500 rpm and L=11.2 Kg

Therefore, POWER = 16.8 x 0.736 =12.3648 Kw. (since 1 metric $hp = 0.736$ Kw)

A-17.5 STANDARDIZATION OF BHP

Since the laboratory condition (site condition) were not the same as standard condition accepted by ISO ,so it needs standardization for finding out standard power and standard fuel consumption .

For standardization of BHP (brake horse power) and BSFC (brake specific fuel consumption), ISO-3046 with corresponding BS 5514 has been used.

From BS 5514 $t_r = 27$ °°, $T_r = 273+27 = 300$ °K and pr = 100 Kpa are taken as standard temperature and pressure.

Where as our laboratory temperature varied from 27.5 °C to 35°C and pressure also varied. Px=100.28 Kpa from sample data.

From the sample data (at the ignition timimg of 15 °BTDC),

tx **= 91.4** $t_w = 83.$ \varnothing _x = 72.1 91.4 83.3 83.3 83.3 71.13 71.13 71.13 71.14. 70.14 and 70.14 % 91.4 91.4 91. 58 83.3 91.76 and 91.76**°**F 83.3 and 83.3 oF

Now from table A-ll (Annex. F of BS 5514), Water vapor pressure (Øx p_{sx}) is found out for $t_{x}=33\,^{\circ}\mathrm{C}$ (91.4 $^{\circ}\mathrm{F}$) and Øx =0.7213, we get 0x pax = 3.6565

 $px- a\emptyset x \ p s x$ Let us now find out Dry Air Pressure Ratio R₁= (------------) $pr - a\varnothing r$ psr from table A-12 (Annex. E of BS 5514). For the value of water vapor pressure \emptyset _x p_{sx} =3.6565 and p_x =100.28 Kpa, we get

 $p_x - \varnothing_x p_{sx}$ Dry Air Pressure Ratio (R_1) = ---------- =0.9869 $p_r - \varnothing$ rpsr

Using standard temperature and room tenperature we get

 $300°K$ $Tr 273+27$ *Rz=* = ------- = ------- = 0.98039 Tx 273+33 306 OK

Again from table A-16 we get the numerical value of m=0.86, n=0.55 and p=0 for natural aspirated gas fueled spark ignition engine.

Now the ratio of indicated power (k) is found out using the formula k = (Rl) *m(Rz)* n (RJ) P $=$ $(R_1)^m(R_3)^n$ since $p=0$ $=(0.9869)^{0.86}(0.98039)^{0.55}$ $= 0.9785$

The power adjustment factor (α) is now calculated from table A-15. For the value of $k = 0.9785$ and mechanical efficiency $\lambda_{\text{Im}}=0.85$, we get $\alpha/2 = 0.976$, which is the power adjustment factor.

The fuel consumption adjustment factor (ß) is similarly found out by interploting the value of k= 0.9785 and m=0.85 in table $A-14$., from where we get the value of $B = 1.00315$.

Brake horse power is now adjusted by using the formula-

 P_x = α Pr where P_r - Brake horse power at standard condition $P_{\bm{x}}$ - Brake horse power at laboratory condition

Hence Standard Power $P_r = P_x / x = (16.8x0.736)/0.976 = 12.678$ Kw

A-17.6 CALCULATION OF BRAKE SPECIFIC FUEL CONSUMPTION. **---**

Brake Specific Fuel Consumption at standard condition is now found out from the relation, $\bar{b_x} = \beta \cdot b_r$, where b_x is Specific Fuel Consumption at room condition and br is Specific Fuel Consumption at standard condition.

Now using sample data ,

bx 1.038x60x60 b_r = $- --$ n 12678 x1.00315

 $b_r = 293.82$. $gm/Kw-hr$

Ilence Brake Specific Fuel Consumption at standard condition is $b_r = 293.82$.gm/Kw-hr.

A-17.7 CALCULATION OF BRAKE SPECIFIC ENERGY CONSUMPTION. ---------------

It is calculated on the basis of Lower Calorific Heating Value = 45 Mj/Kg CNG, the value used by Liquid Fuel Management Group Ltd of Newzealand.

Brake Specific Energy Consumption (BSEC) = BSFCxFuel Value $(kg/Kw-hr \times Mj/Kg)$. \mathbf{r}

 293.82×45 (kg/Kw-hrxMj/kg) Thus, BSEC $=$ 1000

 $= 13.22$ $\frac{Mj}{Kw - hr}$

A-17.8 CALCULATION OF BRAKE THERMAL EFFICIENCY ---~-----------------------------------

> Brake Thermal Output Brake Thermal Efficiency. $=$ Thermal Imput

 $12.678x103(Kwxjoules/sec-Kw)$ B.Th.Eff. = -----------------------------~---------- 1.0345x10-3x45xl06(kg/sec x joules/kg)

 $= 0.2723 = 27.23\%$

A-17.9 CALCULATION OF EQUIVALENCE RATIO.

Theoritical value of air fuel ratio Equivalence ratio $=$ Actual value of air fuel ratio

; .

Let us now find out the theoritical value of air fuel ratio from the combustion equation of natural gas.

The composition of natural gas is as follows as mentioned in appendix table A-3 .

 $\Delta_{\rm{m}}$

A 17.9.1 Combustion of Methane (CH_4) :

 CH_4 + 20₂ = CO₂ + 2H₂O

Therefore, $(12+4)CH_4 + 2(32)O_2 = (12+32) CO_2 + 2(16+2) H_2$.

Hence, (12+4) Kg of Methane combines with 2x32 Kg of Oxygen for complete combustion to give (12+32) Kg of Carbon dioxide and 2x18 Kg of vapor.

Hence 16 Kg. of CH4 needs 64 Kg of Oz for its complete combustion (theoritically).

Again from the gravimetric analysis, air has 77% Nz by weight and 23% Oz by weight. So 1 Kg of Oxygen is present in 100/23 Kg of air.

Therefore 16 Kg of Methane needs 64 Kg of Oxygen which is 100 available in 64x --- Kg of air. 23

Hence one Kg of methane require 64 100 x -- = 17.39 Kg of air. 16 23

Since the compressed natural gas contains 94.35022 % of CH4. Therefore 1 Kg of compressed natural gas needs 17.39xO.9435022 Kg of air = 16.409078 Kg of air for complete combustion of its $CH₄$.

A 17.9.2 Combustion of Ehtane (C2H6): __________________

 $2C_2H_6$ + $7O_2$ = 4 CO_2 + 6 H₂O

2(24+6)Kg of C2H6 + 7(32)Kg of 02 gives 4(12+32) Kg of C02 + 2 (18) Kg of H20. 100 224 x Kg of air. Hence 1 Kg of C2 H6 needs 23 60 100 224 2.56423 Therefore 1 Kg of natural gas needs ----x ----- x -------23 60 100 0.4162228 Kg of air for complete combustion of its C2H6 A 17.9.3 Combustion of Propane (C3 H8): **---------------------** $C_3H_8 + 5 O_2 = 3CO_2 + 4H_2O$ Hence 1 Kg of CNG having 0.42% of propane require (5x32) 100 1.09133 ⁼ 0.1725422 Kg of air. **x ------- x** $\frac{1}{2}$ $(36 + 8)$ 23 100 $\sim 10^{-1}$ A 17.9.4 Combustion of n-Butane (C4H10): **-----------------------** 2 C₄ H₁₀ + 13 O₂ = 8 CO₂ + 10 H₂O Hence 1 Kg of CNG having 1.08266% of n-Butane require 100 1.08266 x 32 x13
--- x ------------------**x -------------------** = 0.1688105 Kg of air. 23 100 x 2 (24 +48)

Thus 1 Kg of CNG needs 16.409078 + 0.4162228 + 0.1725422 + 0.1688105 =17.167356 =17.17 Kg of air.

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Since the composition on natural gas may vary and more over there may be other form of n-Octane ($CsH1s$), so the theoritical air requirement may also vary.

For our calculation we have taken the theoretical air fuel ratio 17.17 . Thus the Equivalence Ratio (E.R.) = $17.17/16.409695$ =1.046 where 16.409695 is the actual air fuel ratio as found in the sample calculation and its data.

83.

17.10 CALCULATION FOR REDUCTION OF ENGINE COMPRESSION RATIO.

Compression ratio is changed by increasing clearence volume by removing metal from piston head or piston cavity. If Vc is the clearence volume of the engine initially. Vs is the swept volume.

 V_c +Vs Then the compression ratio $C_r =$ -------- $V_{\rm C}$

Now, let Vr be the volume of metal to be removed for reducing the compression ratio to a desired value, then final compression ratio becomes

 V_c +Vs+Vr $(V_{ref}) =$ ---------- $V_c + V_r$

Such as an extent

So choosing any compression ratio we may calculate the volume (V_r) to be removed from the piston.

APPENDIX A-1S

86

A-1S.l REDUCTION OF ENGINE COMPRESSION RATIO: **-------------------------------------**

The engine compression ratio was reduced in the following steps :

- a. The diesel engine was properly placed on a support for dismantling of its cylinder head.
- b. The tappet cover from the cylinder head and crank case were removed.
- c. Cylinder head bolts were loosened and both the cylinder head. were removed from the engine block.
- d. Connecting rod's bolts were removed for opening the piston and connecting rod from the crank shaft.
- e. After removing the connecting rod's hubs, the connecting rods with piston assemblies were pushed upward to bring out the pistons and connecting rods through the upper side of the cylinders.
- f. The pistons and connecting rods assemblies were placed on a table.
- g. Piston pin and its lock were removed and piston was seperated from its connecting rod. Piston rings were also removed from the piston.
- h. To reduce compression ratio, either piston head is cut off or piston groove (precombustion chamber) is enlarged to increase clearence volume for reducing the compression ratio to desired value.
- i. In our case the compression ratio was reduced from 17:1 to 12:1. This was done by increasing the clearence volume to the predtermined or calculated value. The calculation for volume to be enlarged has been shown in the sample calculation.

In our laboratory one single cylinder diesel engine (Duetz 210 D, 6.7 Kw) and a double cylinder diesel engine (Duetz 2FL 9121 were modified to hundred percent CNG fueled operation. In the single cylinder diesel engine, the precombustion chamber's diameter (piston cavity diameter)
was enlarged in the machine shop by removing metal from was enlarged in the machine shop by removing the piston on a lathe machine to the calculated diameter

for reducing the compression ratio from $17:1$ to $12:1$. Here the piston head or length of the piston were not changed.

But in the two cylinder diesel engine, an alternative technique was adopted to reduce the compression ratio. In this case the piston head's metal was removed by facing it on a lathe machine to enlarge the clearance volume to the calculated value. The dome inside the precombustion chamber was also removed. The volume of the dome in the piston precombustion chamber was measured by filling the precombustion chamber with water before removing the dome and after removing the dome as measured by micro shringe.

After necessary fitting of the piston and connecting rod to the respective position of the crank shaft, the cylinder head and tapet cover were also fitted to the engine block as they were before.

A-18.2 MODIFICATION AND FITTING OF SPARK PLUGS ON THE CYLINDER HEAD:

The injectors of the diesel engine were removed from the cylinder heads. Two nos. spark plugs of the required specification (as mentioned in table A-2) are selected for fitting them in the place of injectors. If the hole of injector on the cylinder head is smaller than spark plug diameter, then the hole is enlarged in the same angle by drilling to the required diameter and keeping allowence for thread cutting and fitting spark plug to the hole on the cylinder head.

A-18.3 INSTALLATION OF ELECTRONIC IGNITION SYSTEM :

The diesel engine having compression ignition system was modified to spark ignition engine by installing an electronic ignition system with the engine. For doing so, two nos. electronic control units, two nos. induction coils and two nos. spark plugs were installed on the unit along with two other proximity switches as shown in the diagram no. 1.3. The necessary electrical wiring was done as per wiring diagram of electronic ignition system as shown in the schematic diagram no. 1.3.

A-18.4 INSTALLATION OF CNG FUEL SYSTEM BY REPLACING DIESEL FUEL INJECTION SYSTEM

The diesel fuel injection system of the original engine was replaced by CNG fuel system by the installation of pressure regulator, mixer(carburator), CNG cylinder etc as strown Fig.
1.1.1 and Fig. 1.1.2. (5) and Fig. 1.1.2.

